## We claim:

- 1. The process for preparing propylene oxide which comprises
  - a) in a first reaction step reacting hydrogen and oxygen in a solvent/solid catalyst slurry to form hydrogen peroxide, propylene being essentially excluded from the reaction,
  - b) introducing propylene into the hydrogen peroxide and catalyst containing reaction mixture from the first reaction step, and
  - reacting said hydrogen peroxide in a second reaction step with propylene to form propylene oxide.
- 2. The process for preparing propylene oxide which comprises
  - a) in a first reaction zone reacting hydrogen and oxygen in a solvent/solid catalyst slurry to form hydrogen peroxide, propylene being essentially excluded from said zone,
  - b) transferring the hydrogen peroxide and catalyst containing reaction mixture to a second reaction zone, and
  - reacting said hydrogen peroxide in said second reaction zone with propylene to form propylene oxide.
- 3. The process of claim 1 wherein the solid catalyst is a noble metal on TS-1 catalyst.
- 4. The process of claim 3 wherein the noble metal comprises a mixture of noble metals.
- The process of claim 1 wherein the solid catalyst is a palladium on TS-1 catalyst.
- 6. The process of claim 1 wherein the solvent is selected from the group consisting of water, C<sub>1</sub>-C<sub>4</sub> alkanols, carbon dioxide, and mixtures thereof.
- 7. The process of claim 6 wherein the C<sub>1</sub>-C<sub>4</sub> alkanol is methanol.
- 8. The process of claim 2 wherein the solvent is selected from the group consisting of water,  $C_1$ - $C_4$  alkanols, carbon dioxide, and mixtures thereof.
- 9. The process of claim 8 wherein the  $C_1$ - $C_4$  alkanol is methanol.